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A STUDY OF HARTAL SHODHAN ACCORDING TO RAS RATNA SAMUCHCHAY AND ITS PHYSICO-CHEMICAL ANALYSIS

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ABSTRACT

Rasashastra is the branch of ayurveda which deals with the preparation of different medicines using herbals as well as minerals. In this branch there is use of poisonous drugs which are harmful as per the evidence-based medicine. Haratal is most commonly used as medicine besides it harmful or poisonous as per evidence-based medicine. Shodhan is one of the methods which is used before preparation of any formulation. Shodhan deals with the purification of medicines and improvement in their medicinal properties. Shodhan is not only done on minerals but on herbal drugs also. In this present study it is an attempt to evaluate the Shodhan procedure using modern equipments.

INTRODUCTION

Rasashastra is the branch of ayurveda which deals with the preparation of different medicines using herbals as well as minerals. In this branch there is use of poisonous drugs which are harmful as per the evidence-based medicine. Arsenic compounds are being popularly used in ayurveda therapeutics since centuries among them.^[1] Haratal is commonly used in treating diseases like Raktapitta, vatarakta, kushtah and Shleshma rog. [2] Haratal is called orpiment of yellow arsenic with two molecules of arsenic and three molecules of Sulphur (As2S3). Haratal consumed without proper shodhan causes many toxic effects on body. Hence, shodhan of haratal is essential. In Rasashastra there are different methods which used to improve the medicinal properties and overcome the poisonous effects. In that Shodhan is one of the important methods which is used before preparation of any formulation. Shodhan deals with the purification of medicines and improvement in their medicinal properties. It helps to overcome the effects of poison like Tikshnatva, Vyavayi, Vikasi, Ushna. Hartal is used in two forms i.e., shuddhaharatal haratalbhasma. Shuddha Haratal and hartal bhasma is used in many kalpas which are useful in kushthavyadhi, vishamjwara, Vtarakta, Visarpa, Vipadika, Vicharchika, Vrana, Nadivrana, Bhagandara, and Vishamjwara. [3]

AIM OF THE STUDY

A study of Hartal *shodhan* according to Ras ratna samuchchay and its Physico-chemical analysis.

Objective

• To study the concept of *Shodhana*

- Authentication of *Patra Haratal* (Orpiment)
- To perform Shodhana of Haratal by Churnodak.
- Physicochemical analysis of *Haratal* before and after of *Shodhan* procedure.

MATERIALS AND METHOD

Haratal

As a mentioned in *Rasataragini* there are two types of *haratal* i.e., *Pinda haratal* and *Patri haratal*.^[4] As per *Rasatarangini Patra Haratal* is better than *Pinda Haratal*.^[5] *Shodhan* of *haratal* is carried out by *Swedan* in *Churnodak purit dolayantra* method which is mentioned in *Rasasratnasamucchaya*.

For this Procedure two samples of *Ashuddha Haratal* were purchased from local market. after that both samples of *Ashuddha Haratal* were authenticated in certified laboratory. According to the findings Sample no. 2 was selected for the study.

Shodhan of haratal 1) Churnodak Nirman^[6]

Ingredients of churnodaka

- 1. Churna (Lime)
- 2. Water

Procedure

- 1. Course powder of *Churna* (Lime) was taken in quantity of 25gms.
- 2. Then it turned into fine powder form.
- 3. Then powder was put into stainless steel vessel and 6 lit of water was poured into that vessel.

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- The vessel with Churna (Lime) and water was kept untouched for 24 hrs.
- After 24 hrs it was filtered with cotton cloth.

2) Haratal Shodhan^[7]

Ingredients

- Ashuddha Haratal
- Churnodaka

Materials: Stainless steel vessel, Wooden stick, Cottoncloth, Gas Stove, measuring cylinder, Warm water.

Procedure

- 1. AshuddhaHaratal were taken and crushed into small
- The initial weight of haratal was 600gms.
- Those pieces were kept into Cotton cloth and pottali was prepared.

- The pottali was suspended in Dolayantra which containing Churnodakmild heat was given to boil Churnodak.
- 5. When the level of *Churnodak* decreased, again extra churnodak was added.
- This procedure was repeated for 3hrs. After completion of procedure, heating was stopped.
- Then it get cool down pottali was opened and Haratal were washed with warm water and dried it.

Observation during procedure

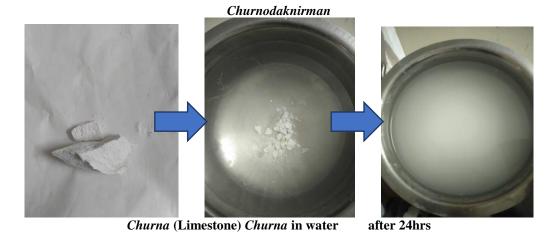
- Fumes liberating from yantra which was having irritating odour.
- Colour of Churnodakwas changed and it gets yellowish. Which was probably a colour of Haratal.
- Haratal becomes soft and brittle.
- Shining and glaze of Haratal was decreased after Shodhan.
- After ShodhanHaratal was weighed and final weight was 575 gms.

Pictures



Sample 1

Sample 2



Haratal Shodhan





Pottali

Churnodak purit dolayantra





In Process

Shuddha Hartal

RESULTS

Table 1: Ashuddha Haratal Organoleptic tests.

Sr	Tests	Results		
		Sample 1	Sample 2	
1	Appearance	Layered, Shiny	Layered, Shiny	
2	Colour	Yellowish orange,	Yellowish	
3	Odour	Characteristic odour of arsenic	Characteristic odour of arsenic	

Table 2: XRF of Ashuddha Haratal.

Sr.	Components	Percentage	
		Sample 1	Sample 2
1	As_2S_3	98.6%	99%
2	Ca	1.38%	1.00%
3	As	60.1%	60.3%
4	S	38.6%	38.7%
5	Fe	-	0.0217%
6	Sb	-	0.0077%

Hence as per above XRF study Sample no 2 was selected because it contains 99% of As₂S₃.

Table 3: Shuddha Hartal Organoleptic.

Sr.	Test	Result
1	Appearance	Layered
2	Colour	Yellowish
3	Odour	Characteristic odour

Table 4: Weight difference table.

Initial weight of Ashuddha Haratal	600gm
Final Weight of Shuddha Haratal	575gms
Total Weight loss	25gms

Table 5: XRF study of ShuddhaHaratal.

Sr.	Elements (mass%)	Percentage	
		Ashuddha haratal	Shuddha hartal
1.	As2S3	99.0%	99.2%
2.	Ca	1.00%	0.517%
3.	As	60.3%	60.4%
4.	S	38.7%	38.8%
5.	Fe	0.0217%	0.310%
6.	Zn	-	0.0098%
7.	Sb	0.0077%	=

DISCUSSION

After shodhanof Haratal by above method, Haratal was subjected to its physicochemical tests. In the organoleptic test it shows the colour and texture difference in shuddha and ashuddha haratal. Shuddha Haratal was yellowish shiny, peculiar odor and crystalline smooth texture. It was observed that the greenish tinge got appeared. The lustre of haratal was reduced in shuddha hartal. During Shodhan of Haratal total 600gms of ashuddha haratal was taken and after shodhan we got 575gms of shuddha haratal Total 25gms loss was there.

In this study the percentage of As_2S_3 in Ashuddha Haratal was 99% and besides that it also containing some percentage of Calcium, Iron and Antimony. After shodhan of Haratal it contains As_2S_3 in 99.2%, Calcium Percentage is reduced. Iron is little bit higher and very few percentages of Zinc was observed in Shuddha Haratal which was not found in ashudh haratal it may be due to use of different vessels during process. The percentage of Antimony was totally absent in Shuddha Haratal it may be due to use of Churnodak for Shodhan purpose.

CONCLUSIONS

Shodhana of Haratala reveals that the importance of Shodhana in pharmaceutical process which is probably responsible for therapeutic importance of haratal. Both Shodhita and Ashodhita Haratala shows different physical and chemical properties. Organoleptic characteristics of samples of *Haratala* possess different colors, touch, odor, and also XRF analysis Shows difference in before and after shodhan. XRF of shodhit haratal shows absent of Sb (Antimony) which is harmful for body. As per the study it can be conclude that process of shodhan can alter the effects as well as overcome the poisonous effect on the body.

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